

## CHE611: Experimental Techniques/Instrumentation for Catalysis Research

<b>Program and Course Code</b>	Chemical Engineering CHE611
<b>Course Title</b>	Experimental Techniques/Instrumentation for Catalysis Research
<b>Credit Hours</b>	3
<b>Instructor</b>	Simo Pehkonen
<b>Contact Information</b>	<a href="mailto:spehkonen@masdar.ac.ae">spehkonen@masdar.ac.ae</a>
<b>Office Hours</b>	TBA
<b>Bulletin Course Description</b>	Fundamental theories of reaction rates. Determination of rate parameters using various analytical techniques such as UV-VIS, GC, GC-MS, FTIR, etc. Analysis of rate data and complex reaction networks. Analytical chemistry of catalytic reactions.
<b>Pre-requisites</b>	CHE503, CHE610, or equivalent
<b>Co-requisites</b>	
<b>Course Objectives (Learning Outcomes of the Course)</b>	<p>After completing this course students will be able to:</p> <ul style="list-style-type: none"> <li>• Measure reaction kinetics using widely available analytical tools</li> <li>• Specify the mechanism, rate expressions, and models needed for heterogeneous reactors.</li> <li>• Synthesize detailed reaction networks for catalytic reactions on solid catalyst surfaces, such as zeolites and TiO<sub>2</sub></li> </ul>

<b>Week</b>	<b>Course Topics and Contents</b>	<b>Readings</b>
1	Review of homogeneous reactions	Class Notes and papers
2	Chemistry and mechanisms of homogeneous reactions	Class Notes and papers
3	Review of analytical equipment to be used	Class Notes
4	Data Analysis and Error Analysis, Lab 1 on homogeneous 2 <sup>nd</sup> order reaction via UV-VIS.	Class Notes
5		Class Notes
6	Homogeneous catalyst experiment, Lab 2	Class Notes
7		Class Notes
8	Reactions on solid surfaces	Class Notes
9	Catalytic chemistry I	Class Notes
10	Heterogeneous catalyst experiment, Lab 3	Class Notes
11	Heterogeneous catalysis and kinetics	Class Notes
12	Heterogeneous catalysis and kinetics II	Class Notes
13	Lab 4, energy related experiment on fuels	Class Notes
14	Theory of Phase Transfer Catalysis	Class Notes
15	Phase-Transfer Catalysis Experiment, Lab 5	Class Notes

<b>Relationship of course objectives to program outcomes</b>	
<b>Program Outcome</b>	Demonstrate appropriate depth and breadth of knowledge that is at the frontier of their disciplines
<b>Program Outcome</b>	Use skills of interdisciplinary scholarship and research to integrate multiple perspectives

<b>Out-of-class assignments</b>	
<b>Homework</b>	Three homework assignments based on the experiments performed in the lab.

<b>Course Grading</b>	
<b>Homework</b>	15%
<b>Labs (and lab reports)</b>	5x15% (five lab experiments every other week)
<b>Class Participation</b>	10%
<b>Total</b>	100 %

<b>Class/Laboratory schedule and Methodology</b>	
<b>Class</b>	The class meets 15 weeks, 1 lecture Every other week, 90 minutes each, laboratory component, 3 hours every other week.
<b>Laboratory</b>	
<b>Teaching and learning methodologies</b>	A combination of white board use, Power-point slide presentation, and interactive class discussions to encourage student participation and enhance the learning.

<b>Course Materials</b>	
<b>Textbooks</b>	None
<b>Recommended Ref. Readings</b>	1. B.C. Gates, J.R. Katzer and G.C.A Schuit; <i>Chemistry of Catalytic Processes</i> , McGraw-Hill (1979)
<b>Instructional material and resources</b>	A course website will be set at the beginning of the semester where all course necessary material will be posted including homework assignments and solutions.